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**New high pressure modifications of BPO<sub>4</sub> and BAsO<sub>4</sub>.** By J. D. MACKENZIE, W. L. ROTH and R. H. WENTORF, *General Electric Research Laboratory, Schenectady, New York, U.S.A.*

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In a recent study of phase transformations in various silica isomorphs, Dachille & Roy (1957) found that at pressures and temperatures up to 35,000 atmospheres and 550 °C., only BeF<sub>2</sub> yielded a dense structure corresponding to coesite-SiO<sub>2</sub>. The compounds BPO<sub>4</sub> and BAsO<sub>4</sub>, analogous to silica as SiSiO<sub>4</sub>, are known only in the high-cristobalite modification (Schulze, 1934), a reported quartz form of BPO<sub>4</sub> being still unconfirmed (Shafer *et al.*, 1956). As comparatively higher pressures and temperatures are now available for phase transformation studies, the preparation of the denser polymorphs of these oxides has been reattempted. This note describes the formation of new dense modifications of BPO<sub>4</sub> and BAsO<sub>4</sub>.

BPO<sub>4</sub> was obtained from Victor Chemical Works and BAsO<sub>4</sub> prepared from aqueous solutions of H<sub>3</sub>BO<sub>3</sub> and As<sub>2</sub>O<sub>5</sub>. Chemical analysis showed the purity of the two oxides to be > 99.5% and spectroscopic analysis revealed the absence of heavy metals. The polycrystalline samples were contained in graphite capsules and subjected to pressures of 85,000 atmospheres at 1000–1200 °C. for about 10 min. The compressed materials were crystalline and optically uniaxial. Repeated attempts have failed to produce single crystals of dimensions > 30 μ. Marked increases over the starting materials in density, by sink-float method, and refractive indices, by optical immersion, were found (Table 1). No transformations were

Table 1. *Unit-cell dimensions, density and refractive indices of dense BPO<sub>4</sub> and BAsO<sub>4</sub>*

	BPO <sub>4</sub>	BAsO <sub>4</sub>
$a_0$ (Å)	7.75 ± 0.02	7.91 ± 0.02
$c_0$ (Å)	9.95 ± 0.02	10.32 ± 0.02
$c_0/a_0$	1.29	1.31
$N$ (molecules/unit cell)	9	9
$\rho_{X\text{-ray}}$ (g.cm. <sup>-3</sup> )	3.05	3.98
$\rho_{\text{obs.}}$ (g.cm. <sup>-3</sup> )	3.05 ± 0.05	3.9 ± 0.2
$\omega$	1.636 ± 0.002	1.740 ± 0.002
$\epsilon$	1.648 ± 0.002	1.748 ± 0.002

observed by heating the dense oxides in air at 100 °C. for 24 hours. The X-ray powder diffraction data for both compounds, obtained from Debye-Scherrer and spectrometer methods, are similar and can be indexed on the basis of a hexagonal unit cell, the dimensions of which are presented in Table 1. The spectrometer data are given in Table 2. The  $c_0/a_0$  ratios are appreciably different from those of  $\alpha$ -quartz (1.10) and coesite (1.73). However, allowing for the differences between the Si-O and B-O,

Table 2. *X-ray powder diffraction data for dense BPO<sub>4</sub> and BAsO<sub>4</sub>*

(Co K $\alpha$ ,  $\lambda = 1.5405$  Å)

BPO <sub>4</sub>		BAsO <sub>4</sub>		$hkl$
$d_0$ (Å)	$I$	$d_0$ (Å)	$I$	
3.895	<i>s</i>	3.964	<i>m</i>	110
3.626	<i>m</i>	3.699	<i>s</i>	111
3.363	<i>vwv</i>			200
3.318	<i>w</i>	3.452	<i>w</i>	003
3.204	<i>vwv</i>			201
3.060	<i>vs</i>	3.145	<i>vs</i>	112
2.519	<i>vwv</i>	2.598	<i>vw</i>	113
2.239	<i>m</i>	2.281	<i>w</i>	300
2.183	<i>vw</i>	2.228	<i>w</i>	301
2.092	<i>m</i>	2.162	<i>m</i>	114
2.040	<i>w</i>	2.087	<i>vw</i>	302
1.936	<i>vwv</i>	1.976	<i>w</i>	220
		1.941	<i>vwv</i>	221
1.852	<i>vwv</i>	1.901	<i>vw</i>	303
1.805	<i>vw</i>			222
1.768	<i>vwv</i>	1.831	<i>w</i>	115
1.671	<i>w</i>			223
1.662	<i>m</i>	1.710	<i>m</i>	304
1.527	<i>w</i>	1.568	<i>vw</i>	224
1.485	<i>vwv</i>	1.530	<i>vw</i>	305
1.448	<i>vw</i>	1.476	<i>vw</i>	411
1.403	<i>w</i>			412
		1.433	<i>w</i>	216
		1.381	<i>vwv</i>	117
1.339	<i>vwv</i>	1.357	<i>vwv</i>	413
1.333	<i>vwv</i>			306
1.280	<i>vwv</i>	1.307	<i>vwv</i>	331
1.260	<i>vw</i>	1.296	<i>vw</i>	226

P-O, and As-O bond lengths, the volumes of individual BPO<sub>4</sub> and BAsO<sub>4</sub> groups are approximately equal to that of 2 SiO<sub>2</sub> in  $\alpha$ -quartz.

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